



without any conversion to inosine monophosphate, was converted to inosine and then to hypoxanthine. It was observed that these changes, which occurred in squid muscle, are almost quantitative. In squid muscle, as compared with carp muscle, two specific phenomena can be observed. The first is the accumulation of adenosine monophosphate. A possible pathway for the deamination of this substance, other than direct deamination, is by a combination of a 5'nucleotide phosphatase with adenosine deaminase. This is supported by the fact that adenosine was converted to inosine very rapidly in crude extracts of squid muscle (Saito, T., and Arai, K., unpublished). The second is dependent upon the degree of splitting of adenosine triphosphate. In carp muscle there is a very rapid splitting as soon as freezing is complete, and at the same time inosine monophosphate accumulates; in squid muscle these specific phenomena are not observed.

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<sup>1</sup> Saito, T., and Arai, K., Nature, 179, 820 (1957).

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## Amino-Acid Composition of Hæmoglobin from Thunnus thynnus

WE have recently crystallized<sup>1</sup> hæmoglobin and myoglobin of the saltwater fish, Palamys sarda and Thunnus thynnus. Some physico-chemical and functional properties of these pigments have been described<sup>3-4</sup>. In this communication we report the amino-acid composition of *Thunnus thymnus* hæmo-globin, as determined by chromatography on an ion-exchange resin : a slightly modified version<sup>5</sup> of the method of Moore and Stein.

The hæmoglobin crystals, obtained from ammonium sulphate, were dissolved in water, dialysed against water and hydrolysed with 6 N hydrochloric acid for 24 hr. The quantity of protein present in each

Table 1. AMINO-ACID COMPOSITION OF CRYSTALLINE Thumnus thynnus HEMOGLOBIN (GM. AMINO-ACID/100 GM. PROTRIN)

Amino- acid	Hydrolysate 1			Hydrolysate 2		Mean	
Aspartic acid Threonine Serine Glutamic acid Proline Glycine Alanine Valine Meth- ionine isoLeu- clue Leucine Tyrosine Phenyl- alanine Histidine Lysine Arginine	11.77  4.67  4.71  6.90  2.45  4.45  9.04  7.01  3.22  5.90  12.74  4.81  6.69  4.04  8.95  5.67  .67	11.00 4.5 4.85 6.89 1.66 4.49 8.35 4.99 6.65 3.98 8.40 5.73	4.47	11.69 4.48 4.51 6.67 2.93 4.38 8.92 8.12 2.37 6.10 11.73 4.73 6.44	$11 \cdot 4 \\ 4 \cdot 12 \\ 4 \cdot 10 \\ 6 \cdot 54 \\ 3 \cdot 87 \\ 8 \cdot 58 \\ 6 \cdot 21 \\ 2 \cdot 51 \\ 5 \cdot 32 \\ 11 \cdot 11 \\ 4 \cdot 37 \\ 5 \cdot 93 \\ 4 \cdot 42 \\ 8 \cdot 97 \\ 5 \cdot 46 \\ \end{array}$	11 -28 4 -04 4 -24 5 -88 3 -82 8 -39 6 -89 3 -59 5 -48 11 -28 4 -16 5 -70 4 -02 8 -75 5 -49 Total	$\begin{array}{c} 11\cdot 18\pm 0\cdot 31\\ 4\cdot 38\pm 0\cdot 09\\ 4\cdot 48\pm 0\cdot 16\\ 6\cdot 67\pm 0\cdot 18\\ 2\cdot 40\pm 0\cdot 28\\ 4\cdot 34\pm 0\cdot 22\\ 8\cdot 49\pm 0\cdot 22\\ 7\cdot 28\pm 0\cdot 35\\ 2\cdot 81\pm 0\cdot 22\\ 7\cdot 28\pm 0\cdot 35\\ 2\cdot 81\pm 0\cdot 22\\ 7\cdot 28\pm 0\cdot 35\\ 2\cdot 81\pm 0\cdot 22\\ 6\cdot 68\pm 0\cdot 19\\ 11\cdot 78\pm 0\cdot 30\\ 4\cdot 63\pm 0\cdot 12\\ 6\cdot 37\pm 0\cdot 19\\ 4\cdot 18\pm 0\cdot 10\\ 8\cdot 64\pm 0\cdot 21\\ 5\cdot 58\pm 0\cdot 18\\ 98\cdot 74\end{array}$

sample used for chromatography was calculated on the basis of the nitrogen content (determined by the micro-Kjeldahl method in quadruplicate), assuming the value 16 for the percentage of nitrogen. The analyses were performed on two different hydrolysates of the same sample of crystals.

The results are given in Table 1. The values for proline were calculated using the formula previously reported<sup>5</sup>, since we did not obtain a good separation of this amino-acid from glutamic acid. No results for cystine are available, since the method used is not appropriate for this amino-acid, but it has been identified as cysteic acid by paper chromatography<sup>2</sup>. This work was aided by a grant from the Rockefeller

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<sup>b</sup> Rossi-Fanelli, A., Cavallini, D., and De Marco, C., Biochim. Biophys. Acta, 17, 377 (1955).

## Absolute Configuration of Enantiomorphic Carbanions involved in the Aldolase and Triose Phosphate Isomerase Reactions

DIHYDROXYACETONE phosphate is one of many biological compounds which contain a carbon atom attached to two identical and two dissimilar groups, that is, a carbon attached to groups a, a, b and d. When one of the identical groups of such an 'Ogston atom'<sup>1</sup> or 'mesocarbon atom'<sup>2</sup> participates in a reaction it is of interest to establish the identity of that group<sup>2-6</sup>. Aldolase and triose phosphate isomerase have been shown to labilize only one of the hydrogens bound to the carbinol-carbon of dihydroxyacetone phosphate, and it has further been demonstrated that a different hydrogen is labilized by each of these enzymes". It is the purpose of this communication to establish the identity of the specific hydrogen