to the behaviour of the bound vitamin B_{12} in liver extracts.

Wijmenga et al.⁵ have pointed out that no method of isolating cyanocobalamin from liver extracts has been described that does not involve a proteolytic step, and that when Smith^{9,10} originally described the isolation of vitamin B_{12} he reported that two pink bands were formed on partition chromatography, one of which appeared to be transformed into the other on treatment with a proteolytic enzyme. Cooley et al.⁶ have suggested that vitamin B_{12} may be present in liver in the form of a cobalichrome in which the cyano group is replaced by a protein or polypeptide moiety. That this may be partially true is suggested by the fact that we have invariably found less cyanocobalamin in liver extracts by the chemical method of Boxer and Rickards¹¹ than by microbiological assay, unless the liver extract has first been subjected to cyanide treatment, when the agreement between the two sets of results is fairly good. This confirms Boxer and Rickards's conclusion that "in liver concentrates, cyanocobalamin usually accounts for only a small fraction of total cobalamins".

By contrast, the cyano group is apparently unaffected when intrinsic factor is added to cyanocobalamin, since the results obtained with such a mixture by the method of Boxer and Rickards was in good agreement with the amount of cyano-cobalamin present. This supports the observation of Wijmenga et al.12, based on a study of absorption spectra, that the eyano group is still present in the cyanocobalamin-intrinsic factor complex.

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June 15.

- ¹ Robinson, F. A., Williams, B. W., and Brown, L. H., J. Pharm. Pharmacol., 4, 27 (1952).
 ² Pitney, W. R., Beard, M. F., and van Loen, E. J., J. Biol. Chem., 207, 143 (1954).

- 207, 143 (1954).
 ³ Gregory, M. E., and Holdsworth, E. S., Nature, 173, 830 (1954).
 ⁴ Ross, G. I. M., Nature, 166, 270 (1950).
 ⁴ Wijmenga, H. G., Veer, W. L. C., and Lens, J., Biochim. Biophys. Acta, 6, 229 (1950).
 ⁶ Cooley, G., Ellis, B., Petrow, V., Beaven, G. H., Holiday, E. R., and Johnson, E. A., J. Pharm. Pharmacol., 3, 271 (1951).
 ⁸ Hausmann, K., and Mulli, K., Lancet, i, 185 (1952).
 ⁸ Prusoff, W. H., Meacham, G. C., Heinle, R. W., and Welch, A. D., Abs. Amer. Chem. Soc., 118th meeting, 27A (1950).
 ⁹ Smith, E., Lester, Nature, 161, 638 (1948).
- ¹⁹ Smith, E. Lester, and Parker, L. F. J., *Biochem. J. Proc.*, **43**, viii (1948).
- ^{(1940).}
 ¹¹ Boxer, G. E., and Rickards, J. C., Arch. Biochem., **30**, 372, 382. 392 (1951); **39**, 281 (1952).
 ¹² Wijmenga, H. G., Thompson, K. W., Stern, K. G., and O'Connell-D. J., Biochim. Biophys. Acta, **13**, 144 (1954).

A Riboflavin-producing Bacterium

RIBOFLAVIN synthesis has been demonstrated in a wide variety of micro-organisms by means of growth studies; these have been adequately reviewed from time to time¹. Among the bacteria, members of various genera have been reported to produce riboflavin in very small amounts2, and none has been reported which exceeds the capacity of Cl. acetobutylicum to produce this vitamin.

During the course of routine work, an air-contaminant was found to produce an intense greenish-

yellow pigment in a medium consisting of casein hydrolysate and beef-extract. This pigment appears to be riboflavin; the compound, like the vitamin, has the following properties: (a) it exhibits an intense greenish-yellow fluorescence in ultra-violet light; (b) the fluorescence is instantaneously discharged on the addition of sodium hydrosulphite; (c) it has absorption maxima at 225, 269, 372 and 445 mµ, as measured in the Beckman spectrophotometer.

The air contaminant was found to be a bacterium. It was isolated in the pure state by the usual methods. On microscopic examination it was found to be coccoid, arranged in irregular groups of sizes $0.8-1.0 \,\mu$ in diameter. It is Gram-negative (observed after 6 hr., 12 hr., 24 hr. and 48 hr.).

The amount of riboflavin produced by this organism was determined by carrying out static and agitated fermentations, using a medium containing casein hydrolysate and beef-extract. Riboflavin production was assayed by the fluorimetric method using a pure riboflavin solution as the standard and found to be of the order of 150-250 µgm./ml., the maximum amount being produced in the reciprocating shaker at 26° C., the yield obtained being 250 µgm./ml. This high yield of riboflavin by a bacterium is very interesting and further studies on the identification of the organism are in progress.

Acknowledgments are made to Dr. S. C. Roy, Dr. S. K. Bose and Mr. P. R. Pal, of the Department of Applied Chemistry, University of Calcutta, for their suggestions during the progress of the work.

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May 13.

Knight, B. C. J. G., Vitamins and Hormones, 3, 108 (1945). Peterson, W. H., and Peterson, M. S., Bact. Revs., 9, 49 (1945). Lanen J. M., and Tanner, F. W., Vitamins and Hormones, 6, 163 (1948). Pridham G. T., Econ. Bot., 6, 185 (1952).
 Tittsler, R. P., and Whittier, S. O., J. Bact., 42, 151 (1941).

Inhibition of D-Amino-acid Oxidase

PEPTIDES containing D-amino-acids are not attacked by p-amino-acid oxidase^{1,2}. We have found that the action of pig kidney D-amino-acid oxidase on DL-leucine was inhibited by pL-leucinamide, pL-leucylglycine, DL-leucylglycylglycine, and by glycyl-DL-leucine. These compounds also inhibited the enzymic oxidation of **DL**-valine and **DL**-phenylalanine. Thus. DL-leucinamide when added to DL-leucine, DL-valine or DL-phenylalanine in a molar ratio 5:1 inhibited the oxidation by 35-45 per cent. DL-Leucylglycine inhibited the oxidation of pL-leucine by 40-45 per cent, of DL-phenylalanine by 50 per cent, and of DL-valine by 60 per cent. DL-Leucylglycylglycine inhibited the oxidation of DL-phenylalanine by 25 per cent, of DL-leucine by 30 per cent, and of DL-valine by 65 per cent. Glycyl-DL-leucine gave similar results. With glycylglycine no inhibition was obtained.

It was also found that derivatives of DL-valine, DL-leucine and DL-phenylalanine in which one of the amino-hydrogens was substituted by ethyl, hydroxyethyl or cyclohexyl radicals also acted as inhibitors of p-amino-acid oxidase. Thus, pL-N-ethylphenylalanine inhibited the oxidation of DL-phenylalanine