

LETTERS TO THE EDITORS

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Masses of Krypton-82, -84 and -86 and Xenon-129 and -132

In the course of our atomic mass measurement programme, we have recently had occasion to measure the masses of some of the isotopes of krypton and xenon by the mass-spectrographic doublet method. These measurements are briefly described below.

Krypton and xenon ions were obtained from a Nier-type of ion source¹ in which gas molecules are ionized by electron bombardment. The ionizing electrons were given energies of about 250 eV. This resulted in a copious production of the doubly-charged krypton and triply-charged xenon ions which were needed for the mass comparisons.

To obtain the mass of xenon-129, the triplet at mass number 43 consisting of singly-charged C₂OH₃ and C₃H₇ and triply-charged ¹²⁹Xe was photographed. The mass of xenon-132 was obtained from the triplet at mass number 44 consisting of singly-charged CO₂ and C₂OH₄ and triply-charged ¹³²Xe. The following packing fraction differences (× 10⁴) were obtained: C₂OH₃ - ¹²⁹Xe, Δf = 11.678 ± 0.03; C₃H₇ - ¹²⁹Xe, Δf = 20.152 ± 0.03; C₂OH₄ - ¹³²Xe, Δf = 13.083 ± 0.03; and CO₂ - ¹³²Xe, Δf = 4.906 ± 0.02. Assuming the masses of ¹H and ¹²C to be 1.008144 ± 0.000008 and 12.003798 ± 0.000025, respectively, the packing fraction of xenon-129 is computed to be -4.230 ± 0.03 from the C₂OH₃ - ¹²⁹Xe comparison, and -4.245 ± 0.04 from the C₃H₇ - ¹²⁹Xe comparison. Similarly, two values for the packing fraction of xenon-132 are found: these are -3.953 ± 0.03 from the C₂OH₄ - ¹³²Xe doublet, and -4.043 ± 0.03 from the CO₂ - ¹³²Xe doublet. The errors which are assigned are probable errors based on the internal consistency of the measured values.

The masses of krypton-82 and -84 were obtained from doublets at mass numbers 41 and 42, the former consisting of singly-charged C₃H₅ and doubly-charged ⁸²Kr, and the latter consisting of singly-charged C₃H₆ and doubly-charged ⁸⁴Kr. The mass of krypton-86 was computed from the triplet at mass number 43, consisting of singly-charged C₂OH₃, doubly-charged ⁸⁶Kr and triply-charged ¹²⁹Xe. The following packing fraction differences (× 10⁴) were found: C₃H₅ - ⁸²Kr, Δf = 20.219 ± 0.03; C₃H₆ - ⁸⁴Kr, Δf = 21.668 ± 0.03; C₂OH₃ - ⁸⁶Kr, Δf = 14.805 ± 0.035; and ¹²⁹Xe - ⁸⁶Kr, Δf = 3.156 ± 0.03. From these data the packing fraction of krypton-82 is computed to be -7.508 ± 0.035, that of krypton-84 to be -7.321 ± 0.035; and the two values for the packing fraction of krypton-86 are -7.357 ± 0.04 from the C₂OH₃ - ⁸⁶Kr comparison, and -7.392 ± 0.04 from the ¹²⁹Xe - ⁸⁶Kr comparison, using as the packing fraction of xenon-129 the value found above.

Since three of these doublets were studied by Aston² in 1937, it is thought that readers of *Nature* may be interested in the comparisons shown in Table 1, in which Aston's values for these doublet separations are tabulated together with our values for the corresponding doublets. The two sets of values are seen to be in excellent agreement, a fact which probably should be regarded as a substantiation of both. We are aware of the conservative errors assigned by Aston to his measurements, and it may

Table 1. COMPARISON BETWEEN PRESENT VALUES AND THOSE OF ASTON (1937)

Doublet	Packing fraction difference × 10 ⁴	
	Aston	Present work
C ₃ H ₅ - ⁸² Kr	20.20 ± 0.15	20.219 ± 0.03
C ₃ H ₆ - ⁸⁴ Kr	21.73 ± 0.15	21.668 ± 0.03
C ₃ H ₇ - ¹²⁹ Xe	20.16 ± 0.1	20.152 ± 0.03

be that the smaller errors associated with the present measurements are more an indication of optimism than of increased precision. Regardless of the relative accuracies, it is a pleasure to see the pioneer work of Aston standing so well the test of time. It will be recalled in this connexion that Aston's value³ for the mass of sulphur-32, although under a shadow for some years, has recently been shown⁴⁻⁷ to be correct.

The new mass values resulting from the work described in this communication are shown in Table 2. Their relation to other current mass measurements will be discussed in the *Reviews of Modern Physics* in an article now in preparation by one of us (H. E. D.).

Table 2. LIST OF NEW MASS VALUES

Nuclide	Packing fraction × 10 ⁴	Mass
⁸² Kr	-7.51 ± 0.04	81.93843 ± 0.00029
⁸⁴ Kr	-7.32 ± 0.04	83.93850 ± 0.00029
⁸⁶ Kr	-7.37 ± 0.03	85.93658 ± 0.00024
¹²⁹ Xe	-4.236 ± 0.02	128.94536 ± 0.00026
¹³² Xe	-4.00 ± 0.05	131.94727 ± 0.00060

The present experiments have been done with a large Dempster-type double-focusing mass spectrograph⁸. The work has been supported by the United States Atomic Energy Commission.

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May 2.

¹ Nier, A. O. C., *Rev. Sci. Instr.*, **11**, 212 (1940).

² Aston, F. W., *Nature*, **140**, 149 (1937).

³ Aston, F. W., *Proc. Roy. Soc., A*, **163**, 391 (1937).

⁴ Smith, L. G., *Phys. Rev.*, **81**, 295 (1951).

⁵ Duckworth, H. E., Stanford, G. S., Olson, J. M., and Kegley, Cort, *Phys. Rev.*, **81**, 236 (1951).

⁶ Nier, A. O., *Phys. Rev.*, **81**, 624 (1951).

⁷ Ewald, H., private communication (April 3, 1951).

⁸ Duckworth, H. E., *Rev. Sci. Instr.*, **21**, 154 (1950).

Measurement of Very Low Velocities of Water-flow

THE need has been felt for some years past for an instrument which will measure low velocities of water-flow over the cross-section of an open channel or a pipe. The use of the Pitot tube, which has proved so valuable in the case of the flow of gases, becomes quite impracticable in the case of slow-moving liquids owing to the difficulty of measuring the extremely small pressure differences involved and also to the need for strict thermal control of the connecting tubes. Rotating-vane flow-meters are likewise impracticable at low flow values, because of uncertain spindle friction. Exposed hot wires accumulate gas bubbles around themselves, and the enclosed type becomes excessively slow in response as the velocity is decreased.

This same instrumental need is specially acute when research is attempted into the laws governing the transport of solid grains by water. Here the require-