

time when the surface of the iron shows a lead-like colour, rapidly oxidize the lead-coloured substance to a brownish one, whereas they have no visible effect on iron not activated. The difference between activated and non-activated iron can, of course, be shown immediately after the activation by the addition of a suitable organic reagent, such as nitroguanidine. However, it can also be shown by the addition of an alkaline copper tartrate solution, which produces a brilliant coating, often in several colours, on the activated metal.

In most cases the concentration of the sodium hydroxide was 2-5 *N*. The activation was carried out either by mixing the iron material (screws, nails, pieces of soft iron wire) with small amounts of zinc or aluminium or by means of an external current. The experiments were made at room temperature. Further details will be published later.

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¹ [Nature, 157, 550 (1946).]

Effect of Unsaturated Fatty Acids on the Acid Production of *Lactobacillus helveticus*

WE have recently described the effect of certain unsaturated fatty acids upon growth and acid production of *Lactobacillus helveticus* and other Gram-positive bacteria^{1,2}. When added to the culture medium in a concentration of about 1:100,000, linolenic, linoleic and oleic acids, in that order of efficiency, inhibit the growth of these bacteria. The inhibition can be reversed by the addition of similar concentrations of certain surface-active agents, such as cholesterol or lecithin. We have suggested² that these phenomena may be of a physico-chemical nature.

In the case of two unsaturated fatty acids we have since tested for inhibitory activity the *trans*-isomers of the naturally occurring *cis*-forms. The *trans*-forms differ from the *cis*- in their molecular structure, length of crystal cell and surface area of monolayer.

Effect of α - and β -elaeostearic acids. It was not possible to test α -elaeostearic acid or its optical isomers by the methods originally described^{1,2}, as these substances did not remain stable during aerobic incubation at 37° C. The method adopted therefore was to dissolve the unsaturated fatty acid in ethanol, to add it with sterile precautions to autoclaved media, and to incubate the tubes anaerobically in carbon dioxide. Under these conditions reproducible results were obtained.

TABLE 1. ACID PRODUCTION OF *L. helv.* UNDER ANAEROBIC CONDITIONS (IN CARBON DIOXIDE).

	0.1 <i>N</i> -acid formed, ml. (24 hours)	0.1 <i>N</i> -acid formed, ml. (48 hours)
10 ml. CHCl ₃ -extracted medium*	4.9	15.6
10 ml. CHCl ₃ -extracted medium + 160 μ gm. linoleic acid	0	0
10 ml. CHCl ₃ -extracted medium + 160 μ gm. α -elaeostearic acid	1.8	10.9
10 ml. CHCl ₃ -extracted medium + 160 μ gm. β -elaeostearic acid	6.1	16.4
10 ml. CHCl ₃ -extracted medium + 160 μ gm. elaidic acid	6.1	15.4
10 ml. CHCl ₃ -extracted medium + 160 μ gm. oleic acid	0	2.0

* The riboflavin-free medium used was improved³ to give higher acid production than that used in experiments in Table 2. Riboflavin was added in amounts of 0.5 μ gm. per tube.

As noted in Table 1, α -elaeostearic acid in concentrations of 160 μ gm. per 10 ml. medium depressed the acid production of *L. helveticus* considerably for 48 hours, whereas the *trans*-isomer, β -elaeostearic acid, was without inhibitory activity. That α -elaeostearic acid was less efficient than linoleic acid may have been due to partial destruction of the former even under anaerobic conditions.

Effect of oleic and elaidic acids. When the lactobacilli were grown aerobically in presence of oleic acid and elaidic acid, respectively, the oleic acid inhibited growth and acid-production completely for the first 24 hr. of the incubation period (Table 2). Elaidic acid (the *trans*-form), however, did not show any inhibiting effect. A similar result was also obtained when both acids were added under the conditions described above, that is, after autoclaving and incubating anaerobically (see Table 1).

TABLE 2. ACID PRODUCTION OF *L. helv.* UNDER AEROBIC CONDITIONS.

	0.1 <i>N</i> -acid formed, ml. (24 hours)	0.1 <i>N</i> -acid formed, ml. (48 hours)
10 ml. CHCl ₃ -extracted medium*	2.8	7.9
10 ml. CHCl ₃ -extracted medium + 160 μ gm. oleic acid	0.4	2.3
10 ml. CHCl ₃ -extracted medium + 160 μ gm. elaidic acid	2.3	8.7

* The riboflavin-free medium used was that of Kodicek and Worden⁴. Riboflavin was added in amounts of 0.5 μ gm. per tube.

In the case of these two pairs of optical isomers, therefore, the *trans*-forms do not inhibit the acid-production of *L. helveticus* under conditions which permit the naturally occurring *cis*-isomers to do so. The behaviour of the *trans*-forms on surfaces would differ from that of the *cis*-forms, and the results obtained in these experiments are therefore in keeping with the physico-chemical hypothesis of the

inhibitory effect², although they do not necessarily exclude a purely chemical interaction.

We are indebted to Prof. T. P. Hilditch for suggesting these experiments and for supplying us with specimens of elaidic, α -elaeostearic (m.p. 46.5° C.; E_1 per cent at 268 $m\mu$ = 1770), and β -elaeostearic (m.p. 71.2° C.) acids. Dr. T. Moore very kindly helped us with the preparation of a second specimen of α -elaeostearic acid (m.p. 46.3° C.) since the original sample deteriorated rapidly as the result of polymerization or oxidation.

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¹ Kodicek, E., and Worden, A. N., *Nature*, 154, 17 (1944).

² Kodicek, E., and Worden, A. N., *Biochem. J.*, 39, 78 (1945).

³ Kodicek, E., in the press.

Alcohol-Soluble Osteogenetic Substance from Bone Marrow

IN an interesting communication in *Nature*¹, Prof. Lacroix has reviewed his investigations of an osteogenetic substance which can be extracted from the bone tissue with alcohol, and for which he proposes the name 'osteogenin'. We should like, however, to point out that this substance has already been reported by Levander², Annersten³ and Bertelsen⁴. We are at present engaged in a more detailed investigation of its chemical nature. In experiments performed in collaboration with Dr. Hans Bohr, we have confirmed Annersten's finding that the osteogenetic substance can be shaken out from its alcoholic solution with benzene. Furthermore, by saponification we have divided the lipid substances in the bone marrow into (a) the unsaponifiable fraction, (b) the fatty acid fraction and (c) the residual solution after saponification and extraction of (a) and (b). The osteogenetic activity in (a) was very low, that in (b) somewhat higher, while fraction (c) was completely inactive. The investigation is being continued.

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¹ *Nature*, 158, 576 (1945).

² *Surg. Gynecol. Obst.*, 67 (1938). *Nature*, 155, 148 (1945).

³ *Acta Chirurg. Scand.*, 84, Suppl. 60 (1940).

⁴ *Acta Orthoped. Scand.*, 15, 139 (1945).

Choline Esterase and its Specificity

A SPECIFIC choline esterase is an enzyme which hydrolyses acetylcholine at a higher rate than any other esters¹. Richards and Cutkomp², however, have found that choline esterase from insect nerves is more active on acetyl- β -methylcholine than on acetylcholine. I have also arrived at similar results with other invertebrates³. Thus the whole animal of *Patella vulgata*, and the ampulla and podia of *Asterias rubens* from the west coast of Sweden, split acetyl- β -methylcholine at a higher rate than acetylcholine. Benzoylcholine is very little split. In contrast to this, *Terebratulina caput serpentis* hydrolyses acetylcholine but neither acetyl- β -methylcholine nor benzoylcholine.

Characteristic of the specific choline esterase is said to be the inhibition of the enzyme by excess of substrate. Moreover, acetyl- β -methylcholine but not benzoylcholine is hydrolysed by it. If we compare the activity-*pS* curves (*pS* is $-\log$ molar concentration of the substrate) with acetylcholine and acetyl- β -methylcholine as substrates, these curves might have the same shape. I have done this with different materials and have found that it is actually the case for the choline esterase of brain and red blood cells. It is not so, however, for the choline esterase in the blood of *Helix pomatia*. Thus hydrolysis is depressed when the substrate concentration is above the optimum when the substrate is acetylcholine but not when it is acetyl- β -methylcholine. The optimum substrate concentration for acetylcholine is about $2.6 \times 10^{-3} M$, for acetyl- β -methylcholine $10^{-7} M$. In another respect also the choline esterase in *Helix* blood seems to be distinguished from the specific esterases in, for example, brain and erythrocytes from higher animals or in other tissues from *Helix*. The hydrolysis of acetylcholine by blood from the edible snail seems to be a zero-order reaction, that of acetyl- β -methylcholine a first-order reaction. Acetylneurine is split at the same rate as acetyl- β -methylcholine, tributyrin at a lower rate, benzoylcholine and ethyl acetate are practically not hydrolysed at all.

Among other tissues investigated from *Helix pomatia*, the dart sac is surprisingly active as regards choline esterase; thus *Q* is 10; for comparison, *Q* for the choline esterase of frog muscle is 0.4-0.6⁴. The activity is very distinct from that of the snail blood. The hydrolysis of acetylcholine by choline esterase in the dart sac is not depressed by high substrate concentrations; thus this enzyme should be called 'pseudo'-choline esterase. Benzoylcholine is not at all or only slightly hydrolysed, while acetyl- β -methylcholine is even somewhat more attacked. Extracts of this tissue do not split acetylneurine and ethyl acetate. The activity against tributyrin, on the other hand, is higher relatively than that of choline esterase in *Helix* blood.

These results will shortly be discussed more fully elsewhere. We may, however, conclude that the last word has not been said regard-

ing the specificity of the acetylcholine esterase (cf. ref. 5). The reactions used to distinguish between a 'true' and a 'pseudo'-choline esterase with the help of two choline esters (acetyl- β -methylcholine and benzoylcholine)* do not tell the complete story about the types of this enzyme.

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- ¹ Nachmansohn, D., and Rothenberg, M. A., *J. Biol. Chem.*, **158**, 653 (1945).
² Richards, A. G., Jr., and Cutkomp, L. K., *J. Cell. Comp. Physiol.*, **28**, 57 (1945).
³ Augustinsson, K.-B., *Acta Physiol. Scand.*, in the press.
⁴ Marnay, A., *C.R. Soc. Biol.*, **126**, 573 (1937).
⁵ Augustinsson, K.-B., *Nature*, **156**, 303 (1945).
⁶ Mendel, B., Mundell, D. B., and Rudney, H., *Biochem. J.*, **37**, 473 (1943).

Excretion of Lead in the Bile

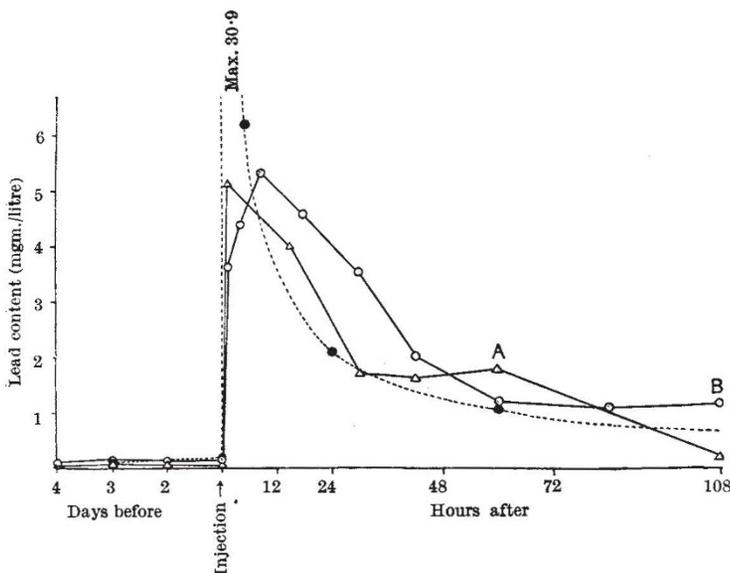
IN the course of experiments by one of us¹, it was found that when single doses of soluble lead salts were injected intravenously in the sheep, only 5-8 per cent of the amount injected was excreted during the five days following injection, most of the lead being retained by the tissues. Of the lead excreted during this period, the greater quantity was found in the faeces as shown in Table 1.

TABLE 1. PERCENTAGE OF EXCRETED LEAD FOUND IN FÆCES AND URINE OF INTACT SHEEP.

Amount of lead injected (mgm.)	Percentage of excreted lead found in	
	Faeces	Urine
10	68.8	31.2
50	74.4	25.6
100	76.7	23.3
200	80.8	19.2
400	83.2	16.8

In these experiments the excretion of lead in the urine was immediate, corresponding with the high plasma-level, but faecal lead excretion was delayed for 4-5 days. These facts suggested that lead was excreted high up in the alimentary tract, for the passage of food residues from colon to anus could scarcely take four days. This was indirectly confirmed by the fact that disburdening of *Nematodirus* sp., a normal inhabitant of the small intestine, occurred at the same time as the peak in faecal lead excretion in one of the experimental sheep. For this reason an experiment was carried out with a sheep in which complete biliary drainage had been accomplished by a modification of Schoregge's method². Following recovery from the operation, bile, urine and faeces were collected every day from this sheep before and after the slow injection of 150 mgm. of lead as the acetate in normal saline. Lead was determined, by a modification of Fischer and Leopoldi's method³, using diphenylthiocarbazon. The concentration of lead in bile and urine is shown in the accompanying graph and the calculated excretion is shown in Table 2.

The graph shows that both biliary and urinary concentrations of lead were enormously increased following the injection, and the close relationship between blood lead and bile and urine lead concentrations will be noted. The curve of concentration of lead in the urine, however, is not smooth, owing to a high value at point A, and afterwards low values. This high concentration appeared due to the onset



LEAD CONCENTRATION IN BILE, BLOOD, AND URINE FOLLOWING INTRAVENOUS INJECTION OF 150 MGm. LEAD.

A, commencement of anuria; B, of polyuria; o—o, bile; Δ — Δ , urine; \bullet \bullet , blood.

TABLE 2. EXCRETION OF LEAD FOLLOWING THE INJECTION OF 150 MGm. LEAD.

Hours after injection	Additional lead found in						Total lead excreted (mgm.)
	Urine		Bile		Fæces		
	mgm.	% of total	mgm.	% of total	mgm.	% of total	
0-24	0.70	15.8	3.87	87.4	—	—	4.43
24-48	0.74	21.3	2.63	76.6	0.09	2.1	3.46
48-72	0.21	23.6	0.68	76.4	0.00	0.0	0.89
72-96	*0.00	0.0	0.49	96.1	0.02	3.9	0.51
96-120	0.06	14.1	0.31	71.4	0.07	14.5	0.44
120-142	0.19	13.4	0.99	69.8	0.24	16.8	1.42

* No urine passed for 24 hours.

of anuria on the fourth day after the injection, while the subsequent low values were associated with an excessive polyuria. The reason for the delay in the maximum for the biliary lead is not clear, but may be related to breakdown of liver cells.

Table 2 shows that the biliary excretion accounted for the larger part of the excretion of lead by the sheep. Over the six-day period, only 7.4 per cent of the injected lead was excreted, and of this amount 17.1 per cent was in the urine, 80.7 per cent in the bile and 2.2 per cent in the faeces. From Table 2 it appears that some excretion does occur through the wall of the gut, and from the time-relationship of the excretion it appears that this is an excretion by the small rather than the large intestine. One interesting observation made was that the normal bile volume per day of 450-500 ml. increased to more than 950 following the injection and then declined to below the normal amount. The sheep died on the seventh day following the injection; serial blood analyses showed a severe anaemia and a considerable nitrogen retention associated with an increase in blood inorganic phosphorus and a fall in blood chlorine. In view of the damage to the kidneys shown by histological sections, a diagnosis of uraemia was made.

The fate of lead injected into the bile-fistula sheep extends the results of the previous experiments on intact animals, shows that in the sheep the bile is the most important single channel for lead excretion and is of interest in view of Anderson's⁴ recent observation on the lead content of biliary calculus.

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- ¹ Blaxter, K. L., unpublished experiments (1945).
² Schoregge, B., *Arch. f. Tierernahrung u. Tierzucht.*, **9**, 722 (1935).
³ Fischer and Leopoldi, *Z. angew. Chem.*, **47**, 90 (1934).
⁴ Anderson, A. B., *Biochem. J.*, **39**, 58 (1945).

Masculinizing Influence of Cystic Ovaries in Female Guinea Pigs

DURING the past few years, about a hundred guinea pigs have been ovariectomized in connexion with a series of experiments on the hormonal control of reproduction. A number of animals had cystic degeneration of both ovaries, and it was noticed that these particular guinea pigs were physically very similar to the male of the species. On the basis of this observation, it was afterwards possible to predict a cystic condition of the ovaries from the external appearance of the animals—in fact, the attendant responsible for their welfare could, latterly, confidently prophesy the operation findings. The pathognomic feature was the altered thoracic contour, particularly the greater anteroposterior diameter, resulting in a postural change in virtue of which these females resembled the male. There was, in addition, hypertrophy of the clitoris, and their weight exceeded that of the average female.

All these animals had been observed for some time prior to operation, had had fairly regular oestrous periods, and the cytology of the vaginal smears showed only minor abnormalities.

The post-operative history of the affected animals and their later reactions to hormonal treatment differed in no way from that of the spayed normal females.

There was no record of abnormal sexual behaviour before operation. Young, Dempsey, Hagquist and Boling¹, who have made an extensive study of 'mounting activity' in guinea pigs, relate this type of sexual behaviour to the maturation of more than the average number of follicles and not to a cystic degeneration of the ovaries, which they find, on the contrary, associated with a total absence of heat. Nevertheless, bilateral cystic degeneration of the ovaries