

nearly. Consequently, light of wave-length greater than λ 1090 Å. should not be able to excite the β -bands. Excitation can be brought about by λ 1200 Å.; if $h\nu_1$ is taken to be in the neighbourhood of λ 2750 Å. (Dutta), $h\nu_2$ comes out as near λ 1250 Å.

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¹ Wulf and Melvin, *Phys. Rev.*, **39**, 180; 1932.

² Dutta, *Proc. Roy. Soc. A*, **138**, 84; 1932.

³ Sen-Gupta, *Bull. Acad. Sci. U.P.*, **3**, 197; 1934.

⁴ Henry, *NATURE*, **134**, 498; 1934.

⁵ Herzberg and Spomer, *Z. Phys.*, **26**, 1; 1934.

⁶ Sen-Gupta, *Z. Phys.*, **88**, 647; 1934.

⁷ Sen-Gupta, *Proc. Roy. Soc. A*, **146**, 824; 1934.

⁸ Jenkins, Barton and Mulliken, *Phys. Rev.*, **30**, 150; 1927.

Electron Diffraction by Vitreous Silica Powder

I HAVE succeeded in showing recently that the comparatively coarse particles of any powder are suitable for structure investigation by means of electron diffraction¹. For this purpose a mineral or other substance is ground carefully in a small mortar. The powder obtained is placed in a sedimentation air tube like that described by Gonell². When blowing the slow jet of air through the heap of powder disposed at the bottom of this sedimentation tube, only the coarsest particles will return, while the minute ones will move with the air blast to the top of the tube where there is a big receiver. By placing there a suitable holder, for example, a loop of wire with a thin celluloid film for the transmission method, or a support for the reflection method, one can collect the particles, the sizes of which depend on the speed of the air blast. It is possible also to collect the smallest particles at the bottom of the tube. For this purpose it is necessary to blow the air for a few minutes only until the smoke appears at the top of the tube, and then to wait, in the case of a tube of one metre in length, in general about 10-15 minutes. Then all particles larger than $0.1-1.0\mu$ in size are precipitated at the bottom of the tube, the remaining being still suspended in the air. To collect the latter on the holder it is sufficient to wait in the case mentioned above about half an hour, after which the process had to be repeated a few times to accumulate a layer of sufficient thickness. It may be noted that the diffraction haloes from the celluloid film disappear completely only when the layer is thick enough. In this way I have obtained sometimes sufficiently clear diffraction patterns from specimens, which were otherwise quite unsuitable for electron diffraction, for example, ignited magnesia, sand, cements and tripolites. Mrs. L. I. Tatarinowa has proposed another method for preparing the specimens by drying on the celluloid film a drop of liquid with a fine suspension of the substance to be studied. This method, which gives sometimes excellent specimens for electron diffraction, has an especial advantage in that it requires only the minutest portion of substance. I have used successfully both these methods.

Here I will confine myself to the case of electron diffraction by vitreous silica glass, which does not give sharp rings when exposed to X-rays. It is of interest that such an amorphous body behaves like a true crystalline aggregate (see Fig. 1). On

the best photographic plates, I have succeeded in measuring sometimes up to ten and even more distinct rings, which correspond approximately to the following Bragg spacings d :

Int.	hkl	(observed) d	(calculated) d
Weak	—	4.40	—
Very strong	111	4.03	4.04
Strong	002	3.64	3.64
Very strong	022	2.52	2.50
Weak	113	2.17	2.17
"	222	1.99	2.02
"	133	1.65	1.62
Fair	224	1.47	1.46
Weak	242	1.42	1.42
Fair	044	1.26	1.25
Very weak	026	1.11	1.14
" "	046	1.01	0.99

(The first extra ring belongs presumably to an adsorbed organic substance from the vacuum grease.) Some of these rings, namely, the most intense ones, have been observed by L. R. Maxwell and V. M. Mosley³ in their work on electron diffraction by transmission of electrons through thin films of silica glass. In spite of the lack of accuracy of the measure-

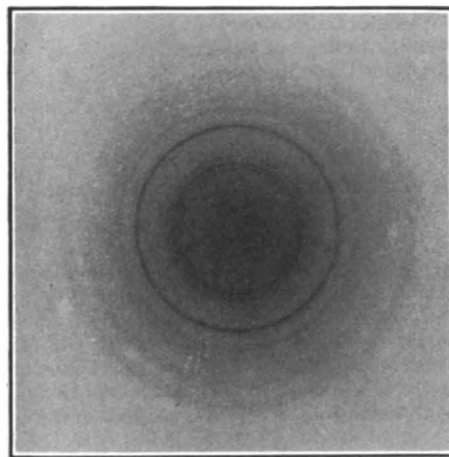


Fig. 1. Electron diffraction pattern of vitreous silica.
 $\times 2$.

ments, I am inclined to think that the above results can only be interpreted as a scattering by tetragonal cristobalite crystallites constituting the vitreous SiO_2 . The axial ratio is 1.06 ; $a=6.87$; $c=7.28$. The size of these crystallites, which is probably of the order of $15-20$ Å., as estimated by J. T. Randall, H. P. Rooksby and B. S. Cooper⁴ by measuring the breadth of X-ray diffraction band, is sufficient to give very sharp diffraction rings. The electron wavelengths in my experiments were about 0.06 Å., while these authors, as well as B. E. Warren⁵, used an X-ray wave-length of 1.54 Å., so that the sharp electron diffraction patterns with many lines are, of course, the consequence of greater resolving power of the electron rays. Hence the method of electron diffraction is applicable to a much larger range of substances than hitherto supposed.

A fuller account is being published elsewhere.⁶
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¹ *C. R. Acad. Sci. U.S.S.R.*, No. 7-8, 461, March 1935.

² *Tonindustrie Z.*, No. 13, 243; 1929.

³ *Phys. Rev.*, **47**, 331; 1935.

⁴ *Z. Krist.*, **75**, 196; 1930.

⁵ *Phys. Rev.*, **45**, 657; 1934.

⁶ *Z. techn. Phys.* (Russian).