composed of 4 hydrogen atoms and 2 negative electrons, then vanadium should have 51 hydrogen atoms and somewhere about 34 negative electrons.

The elements beryllium, neon, magnesium, silicon, chlorine, and argon do not seem to come properly in the curve. If their atomic weights have been correctly determined, then there must be something peculiar about these elements.

I have also drawn a curve from the atomic weights as given in Bloxam's "Chemistry," hydrogen being taken as 1, and have produced a similar curve to the one given here, except that the latter part does not rise so steeply. S. G. BROWN.

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Chemical Warfare and Scientific Workers.

LIKE Prof. Soddy (NATURE, November 4, p. 310), I have received an invitation from the War Office to become an associate member of the "peace" organisation which is to "develop to the utmost extent the offensive and defensive aspects of chemical warfare." I have had enough practical experience of the experimental side of chemical warfare to know what it involves, and I have without any hesitation refused to join the new Committee.

In the first place, the project is simply wicked. Education stands for something more than the acquisition of knowledge, and if at the present time I lent any support to the activities of the Committee I feel that I should necessarily be quite unfit to take any part in the training of young minds. To do what I can to promote in everyone the faith that war is done with has become part of my business because it is the world's business. In the second place, the project is futile. No real progress will be made in discovering new methods of offensive chemical warfare except by people who have their heart in it; perfunctory adhesion to an official organisation will discover nothing worth knowing.

Is any intelligent person—and only intelligent people would be of any use in this very complicated subject at this point in the world's progress going really to put his heart into the search for methods of killing other people? I think not, even in the case of professional soldiers. Some may comfort themselves with the idea that they will escape the moral difficulty by engaging only with defensive methods. This will be equally futile, for adequate defence can only follow discoveries on the offensive side; it cannot precede them. It is impossible to devise protection against offensive agents which are unknown, just as on the medical side it is impossible to work out methods for the cure of lesions of an unknown nature. The only effective preventions and cures which can be prevised are ethical, and a War Office Committee is not quite the best atmosphere for that.

It may be extravagant to expect that all civilians will refuse to support this part of our "peace organisation," but I hope they will. A. E. Boycorr.

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October 5.

Testing Einstein's Shift of Spectral Lines.

A word of caution may not be amiss in respect of the suggestion made by Sir Oliver Lodge in NATURE of October 28, p. 280. The rotational stresses in the disc, though very large, may not portend immediate dissolution in steel, but what of the glass (?) of the vacuum tubes? The stresses, like the gravitational effect, increase as the square of the angular velocity. The method would seem well calculated to develop pleochroic effects in glass. CHARLES CHREE.

October 30.

NO. 2663, VOL. 106

Contractile Vacuoles.

CONTRACTILE vacuoles are found only in those cells which lack a continuous cell-wall. This appeared to suggest that the function of the contractile vacuole is to eliminate dissolved crystalloids and so to keep down the osmotic pressure distending the semipermeable protoplasm. Otherwise the latter, lacking the support which a continuous cell-wall gives, would continue to stretch and would finally rupture.

There is, however, another, and possibly more plausible, point of view, namely, that the contractile vacuole is, in point of fact, this rupture. Suppose a small accumulation of a soluble crystalloid in a semi-permeable gel which exhibits slight elasticity and slight tenacity—qualities which the protoplasm of the cell appears to possess. The osmotic pressure of the crystalloid will push back the protoplasm, overcoming its rigidity. Thus a cavity is formed which enlarges as water flows into it. Expansion will proceed until but a very thin film of protoplasm separates the solution from the surrounding water. Later expansion causes continued thinning of the film until its tenacity suddenly gives way, and the solution contained in the vacuole becomes, through the rupture, continuous with the surrounding water. The elasticity of the protoplasm now asserts itself, and the walls of the cavity are driven together. The semi-fluid, viscid—constituents of the protoplasm secure the healing up of the rupture and the obliteration of the cavity, while the viscosity of the surrounding substance leads to a delay in recovery marked by the appearance of the radiating canals.

Thus we may look upon a contractile vacuole, not as an organ of a cell, but rather as the effect of the local accumulation of any soluble substance. In fresh-water naked protoplasmic organisms the formation of a cavity surrounding this accumulation and the periodic forcible ejection of scme of the solution are rendered inevitable by the physical properties of protoplasm. When once a cell acquires a complete cell-wall, the protoplasmic film receives sufficient support and the vacuoles become permanent.

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School of Botany, Trinity College, Dublin, October 22.

Visibility of the Landscape during Rain.

On a recent visit to the mountains of North Wales the writer was impressed with the variations in the visibility of the landscape when rain was falling. In the lower valleys a storm which may be sufficient to wet thick clothing through in a few moments may leave the contours of the mountains quite distinct at several miles distance. On the other hand, a mountain drizzle or "Scotch mist" may render everything invisible at a few yards. An elementary treatment of the subject brings out one or two points of interest.

Let it be assumed that the raindrops are perfectly opaque and that the atmosphere is otherwise perfectly transparent, both assumptions being, in general, close approximations to the actual state of affairs.

Consider a heavy rainstorm during which rain falls at the rate of 1 cm. per hour, or 0.00028 cm. per second. The raindrops appear to be most often of 1 to 2 mm. diameter. Taking the lower value (1 mm.), the volume of the drop is 0.5×10^{-3} c.c. According to Stokes's law

$$v = \frac{2r^2(\rho - \rho_1)g}{9\eta},$$

and at 15° C., η , the viscosity of air, is 181×10^{-6} , so that v = 3000 cm. per sec.

Consequently, the depth of water which falls in