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Single cell dynamics and nitrogen transformations in the chain forming diatom *Chaetoceros affinis*

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Colony formation in phytoplankton is often considered a disadvantage during nutrient limitation in aquatic systems. Using stable isotopic tracers combined with secondary ion mass spectrometry (SIMS), we unravel cell-specific activities of a chain-forming diatom and interactions with attached bacteria. The uptake of ¹³C-bicarbonate and ¹⁵N-nitrate or ¹⁵N-ammonium was studied in *Chaetoceros affinis* during the stationary growth phase. Low cell-to-cell variance of ¹³C-bicarbonate and ¹⁵N-nitrate assimilation within diatom chains prevailed during the early stationary phase. Up to 5% of freshly assimilated ¹³C and ¹⁵N was detected in attached bacteria within 12 h and supported bacterial C- and N-growth rates up to 0.026 h⁻¹. During the mid-stationary phase, diatom chain-length decreased and ¹³C and ¹⁵N-nitrate assimilation was significantly higher in solitary cells as compared to that in chain cells. During the late stationary phase, nitrate assimilation ceased and ammonium assimilation balanced C fixation. At this stage, we observed highly active cells neighboring inactive cells within the same chain. In N-limited regimes, bacterial remineralization of N and the short diffusion distance between neighbors in chains may support surviving cells. This combination of “microbial gardening” and nutrient transfer within diatom chains represents a strategy which challenges current paradigms of nutrient fluxes in plankton communities.

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INTRODUCTION

Diatoms are responsible for ca. 20% of global primary production [1], and often produce chain-like colonies in plankton communities. Colony formation in phytoplankton occurs across many classes and impacts both nutrient fluxes and grazing potential [2–5]. Diatom chains are often found when nitrate (NO₃⁻) concentrations and turbulence are high during spring blooms and in upwelling regions [4]. Models, laboratory, and field studies have demonstrated that long chains benefit from increased nutrient fluxes from high ambient water turbulence compared to short chains or solitary cells [6–10]. However, chain formation is considered a disadvantage during stagnant conditions and nutrient limitation due to close proximity between cells leading to within-chain competition for nutrients diffusing to the chain compared to solitary cells [6]. Culture studies show decreased chain-length in the stationary phase when concentrations of NO₃⁻ are low [11], which may be an adaptation to diffusion-limited nutrient availability [4].

Over a decade ago, secondary ion mass spectrometry (SIMS) was introduced to microbial ecology [12]. In combination with stable isotopic tracer experiments, SIMS can reveal cell specific biological activities: carbon (C), dissolved inorganic nitrogen (DIN) assimilation, and transfer in mixed field populations, within colonies, between phytoplankton, and attached bacteria [10, 13–18]. With this experimental tool, it was demonstrated that field populations of chain-forming diatoms, such as *Chaetoceros affinis* can contribute up to 20% of total primary

production despite only representing a few percent of the phytoplankton biomass in late summer when DIN concentrations reach the nM range [19]. Primary production was mainly based on remineralized ammonium (NH₄⁺) (ref. [19, 20]). Interactions between *Chaetoceros* and bacteria in the phycosphere, i.e., the boundary layer with concentration gradients of nutrients and solutes adjacent to the cell surface [21–23] resulted in NH₄⁺ assimilation rates up to four times higher in diatoms than those explained by diffusion from the ambient water [19]. Hence, bacteria-algae interactions can lead to NH₄⁺ production and assimilation within the phycosphere which alleviates diffusion-limited nitrogen (N) regimes in field communities. Transfer of diatom derived C and N to heterotrophic bacteria attached directly to suspended diatoms [15] and to diatoms in aggregates [24] have been shown to be more efficient than to free-living bacteria. How these transfers quantitatively benefit living diatoms at a single cell level in chains as compared to solitary cells has to our knowledge not yet been investigated.

Several *Chaetoceros* species lack high affinity transporters for NO₃⁻ (ref. [25, 26]) and the switch from NO₃⁻ to NH₄⁺ assimilation is a well-known adaptation to regenerated production during summer [27]. Culture studies have demonstrated that xenic diatom strains can survive longer than their axenic counterparts during the stationary phase, presumably due to bacterial remineralization of dead diatoms and release of NH₄⁺ (ref. [27]). NH₄⁺ is key in several processes during the stationary phase which are important for the diatom life cycle and their survival on a

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larger time scale. Sexual reproduction in centric diatoms which is essential to restoring cell size after numerous vegetative cell divisions can be triggered by NH_4^+ (ref. [28]). Formation of sinking diatom aggregates (containing *Chaetoceros*) [29] with microenvironments of NH_4^+ concentrations up to 100-fold higher than in the surrounding ambient water occurs after diatom blooms [10]. Such NH_4^+ microenvironments may promote sexual reproduction between different clones [30]. The high aggregate sinking velocities transport cells to sediments where they can survive as resting stages for decades up to millennia, in dark and anoxic conditions [31, 32]. Resting stages of the chain forming diatom, *Skeletonema marinoi*, assimilate NH_4^+ at low rates in such conditions [33].

Here, we investigated C and N dynamics of *C. affinis* and its associated bacteria at a single cell level to quantify changes in nutrient acquisition and chain length during two weeks of the stationary growth phase, and how *C. affinis* stimulates growth of attached bacteria which can benefit diatom survival on short and long time scales. We report the pattern of cell decay within chains and solitary cells and propose that microbially mediated nutrient transfer between neighboring cells in chains during the late stationary phase can be significant in N limited regimes.

METHOD

Culture conditions and growth

C. affinis (CCAP 1010/27, isolated by Di Prisco, 2008), selected from the Gothenburg University Marine Algal Culture Collection was grown in: N limited F/8 medium mixed from autoclaved sea water [34], at 16 °C, 26 PSU, 12:12 h light:dark (L:D) cycle 60 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ measured using a Scalar PAR Irradiance Sensor (Biospherical instruments Inc. San Diego, CA USA) without mechanical aeration. An inoculum of cells was transferred to 1 L Duran bottles filled with 800 mL modified F/8 medium with 117 $\mu\text{M NO}_3^-$, rather than 220 $\mu\text{M NO}_3^-$ in standard F/8 medium [34], for a final concentration of 3000 cells mL^{-1} determined using a Sedgewick Rafter counting chamber (see Diatom abundance and chain-length). The dissolved inorganic carbon to nitrogen ratio (DIC:DIN) was estimated to 17, resulting in a likely N limited medium. Diatom growth was monitored by a Varioskan Flash multimode fluorescence plate-reader (ThermoScientific Waltham, MA USA) using 425 nm excitation and 680 nm detection wavelength [35]. Fluorescence per cell was calibrated against counting chamber observations over the experiment to estimate PON. Dissolved NO_x^- ($\text{NO}_2^- + \text{NO}_3^-$) concentration was monitored using colorimetric analysis by VCl_3 (ref. [36]). A pilot study was conducted prior to reveal the growth dynamics of the diatom strain and identify when the cultures entered the stationary growth phase as described above (data not shown). The sampling times was selected using the plate reader data.

Experimental setup and isotopic enrichment

We conducted isotopic tracer experiments at day 9 (early stationary phase), day 15 (mid stationary phase), and day 21 (late stationary phase). For each experiment, nine bottles were enriched with 2.6–12% $^{15}\text{N-NO}_3^-$ and 11–13% $^{13}\text{C-HCO}_3^-$ just before the light phase. Triplicates of labeled bottles and an unlabeled control bottle were destructively sampled at each time point (T0, after 12 h in light (L), and after 24 h, 12:12 h L:D), in each growth phase. Samples were taken for isotopic analysis at the single cell level (SIMS/nanoSIMS), bacterial enumeration (DAPI staining), isotopic analysis of particulate organic carbon, and nitrogen (POC and PON), dissolved inorganic nutrients (NO_x^- , NH_4^+ , silicate, ortho-phosphate), and cell counts (fixation with Lugol's solution). In the late stationary phase, nine additional flasks were enriched with 1.0–2.4% $^{15}\text{N-NH}_4^+$ and 12% $^{13}\text{C-HCO}_3^-$ then incubated in light. Triplicate bottles were sampled immediately (T0), after 1.5 h, and 4.15 h after tracer addition along with unlabeled control bottles.

Diatom abundance and chain-length

Samples (1 mL) from each bottle was fixed with Lugol solution and stored at 4 °C until analysis. Cells were counted using a Sedgewick Rafter counting chamber, samples were diluted to achieve no more than 30 cells per square. All samples were transferred using an autoclaved pipette tip with

the end cut off, to avoid breaking chains. Over 300 cells were enumerated and had their chain-length determined for each replicate (over 14,400 counted cells in total).

Dissolved inorganic carbon and nutrients

DIC was determined using Apollo SciTech AS-C5 DIC analyzer, Newark, Delaware, USA, from 12 mL samples stored in exetainers, each sample was measured 3–5 times and calibrated against certified reference material (Scripps Institution of Oceanography of the University of California, USA, Batch #189). Spectrophotometric analyses of dissolved NO_x^- (see above), silicate [37], and phosphate [38] was performed on 1 mL filtered (0.2 μm cellulose-acetate) samples from each time point. Total NH_4^+ concentrations at each time point were determined fluorometrically on 40 mL of filtered sample after 4.15 h in the light [39].

Dissolved and particulate isotope analyses

Samples (10–100 mL) for POC and PON were filtered on to pre-combusted GF/F filters (Whatman, 0.7 μm), dried overnight (60 °C), exposed to 20% HCl fume in a desiccator for 24 h to remove carbonates, and packed in tin capsules (Elementar Analysensysteme GmbH) for analysis in an elemental analyzer coupled to an isotope ratio mass spectrometer (EA-IRMS; ANCA-GSL; Sercon Ltd., Crewe, UK). POC and PON assimilation during 12 h L and 12:12 h L:D was calculated by subtracting the T0 POC/PON from the values after 12 h and 24 h, respectively. POC and PON assimilation during 12 h D was calculated by subtracting 12 h L from 12:12 h L:D. Concentrations of $^{15}\text{NH}_4^+$, $^{15}\text{NO}_2^-$, and $^{15}\text{NO}_3^-$ were determined in all experiments. The $^{15}\text{N NH}_4^+$ pool was determined by conversion to N_2 gas using alkaline hypobromite iodine [40] and measured on a gas chromatograph coupled to an IRMS (GC-IRMS; Thermo Delta V Plus) [41]. Any $^{15}\text{NO}_2^-$ was converted to N_2 gas by reduction with sulfamic acid [42] and measured on GC-IRMS. The $^{15}\text{N-NO}_3^-$ pool was measured by first treating samples with sulfamic acid to remove any NO_2^- followed by reduction of NO_3^- to N_2 with cadmium [42, 43] and measurement on GC-IRMS. No significant nitrification (production of $^{15}\text{NO}_2^-$ or $^{15}\text{NO}_3^-$ in incubations with $^{15}\text{NH}_4^+$) was detected (data not shown).

Secondary ion mass spectrometry

Samples (5 mL) were fixed in PFA (Paraformaldehyde, Electron Microscopy Sciences), final concentration 1–2% and stored at 4 °C overnight. The sample was then filtered onto a TTPP (2 μm pore size) filter and rinsed with PBS buffer (10 \times , pH 7.4). Filters were stored dry at room temperature until analysis. N and C assimilation in individual cells were analyzed using a large-geometry Secondary Ion Mass Spectrometer IMS 1280 (CAMECA, Gennevilliers, France) at the NordSIMS facility (Natural History Museum, Stockholm, Sweden). Sample filters were sputtered with a thin (ca. 5 nm) Au layer, 237 areas of interest were pre-sputtered over 80 \times 80 μm for 180 s with a 10 nA Cs^+ primary beam to remove the Au coating and penetrate the Si frustule. The primary beam was then reduced to ca. 50 pA, which corresponds to a spatial resolution of ca. 0.5 μm , and rastered over a 70 \times 70 μm area for data acquisition. Secondary ions were steered onto the ion optic axis using the Dynamical Transfer Optical System and a mass resolving power of 12,000 $\text{M}/\Delta\text{M}$ was used to separate the CN^- species from potential interferences (e.g., $^{11}\text{B}^{16}\text{O}^-$ at nominal mass 27). The ion species $^{12}\text{C}^{14}\text{N}^-$, $^{13}\text{C}^{14}\text{N}^-$, and $^{12}\text{C}^{15}\text{N}^-$ were measured in peak hopping mode using a low-noise ion counting electron multiplier with an electronically gated 44 ns deadtime, each cycle comprising integration times of 1, 2, and 5 s respectively for a total of 60 cycles. The acquired ion images were processed using CAMECA's Winimage2 software. Regions of interest (ROI) were selected from $^{12}\text{C}^{14}\text{N}^-$ counts (cts) after 12 h L (light) and 12:12 h L:D (light:dark) for all experiments (713 cells), within which the isotope ratios $^{13}\text{C}/^{12}\text{C}$ and $^{15}\text{N}/^{14}\text{N}$ were calculated, including on pixel correction for the detector deadtime. The natural background ratios were determined from the average ratios in the non-labeled controls ($^{15}\text{N}/^{14}\text{N}$ ratio: 0.00366, $^{13}\text{C}/^{12}\text{C}$ ratio: 0.01091).

Bacterial abundance and nanoSIMS

From each flask, triplicates 2 mL samples were fixed in 1–2% PFA as above. From each replicate, 250 μL was filtered onto a black 0.2 μm cyclopore filter. DAPI staining was performed according to ref. [44]. Bacteria per diatom were counted (12,535 bacterial cells counted attached to 477 diatoms). Samples were cut from the same TTPP filters as used for SIMS analysis, gold sputtered, and analyzed using a NanoSIMS 50 L (CAMECA, Gennevilliers, France) at the chemical imaging infrastructure (Astra Zeneca, Mölndal, Sweden). Diatom attached bacteria were analyzed by locating diatoms using the attached light microscope to select the region before

sputtering using a primary 150 pA Cs⁺ beam for 75 s. Free living bacteria were located on the filter by using the live view of secondary electrons in the pre-sputtered area. The area was sampled if bacteria were observed as judged by an increase in ¹²C¹⁴N⁻ abundance during live view. An area of 30 × 30 μm was analyzed with 512 × 512 pixels resolution using a cesium beam with ~18 nA current for 1 ms per pixel for up to 15 planes. Each layer was manually observed for burn-through. ¹²C¹³C⁻, ¹²C¹²C⁻, ¹²C¹⁴N⁻, ¹²C¹⁵N⁻, ²⁸Si⁻, ³²S, and secondary electrons was measured. The mass resolution for ¹³C¹²C and ¹³C₂ was >9000 M/ΔM, ¹²C¹⁴N was >8000 M/ΔM, and >2000 M/ΔM for ¹²C¹⁵N respectively. The ROIs for individual bacteria attached to *C. affinis* and free-living bacteria found on the filter were determined using drift corrected and accumulated ¹²C¹⁴N⁻ cts. The positions of diatoms were verified using ²⁸Si cts and secondary electrons. The isotopic ratio of ¹⁵N:¹⁴N was calculated for ROIs by dividing ¹²C¹⁵N⁻ abundance with ¹²C¹⁴N⁻ abundance. The ¹³C:¹²C isotopic ratio was calculated according to Eq. 1 (ref. [45]).

$$^{13}\text{C}:^{12}\text{C IR}_{\text{bacteria}} = \frac{^{12}\text{C}^{13}\text{C}}{^{12}\text{C}^{12}\text{C} \times 2} \quad (1)$$

Bacterial ¹³C:¹²C and ¹⁵N:¹⁴N ratios were measured until a stable mean and standard error of the mean (SE) was established. Natural background ratios were determined from ratios in the non-labeled controls (¹⁵N:¹⁴N ratio: 0.00373, ¹³C:¹²C ratio: 0.01094).

C and N assimilation and growth

The isotopic ratios from SIMS and nanoSIMS were used as IR_{t1} to calculate k, the C-specific C or N-specific N assimilation rate (h⁻¹) against the control

IR_{t0} (ref. 10):

$$k(h^{-1}) = \frac{IR_{t1} - IR_{t0}}{F_{\text{ambient}}(t_1 - t_0)} \quad (2)$$

where F_{ambient} is the isotopic labeled fraction of either N or C in the ambient water and IR is the isotopic ratio in either bacteria, diatoms, or POM during t_1 (12 or 24 h, NH₄⁺ was extrapolated from 4 h).

The assimilation during incubation was calculated according to:

$$a = k \times t \times M \quad (3)$$

where a was assimilation (g) and M is the C or N mass per cell (g). The time (t) was either 12, 24 h (¹⁵NO₃⁻ experiments), or 4.15 h (¹⁵NH₄⁺ experiments) to normalize incubation time and to compare NH₄⁺ incubations against NO₃⁻. C and N specific growth rates GR_{bacteria} (h⁻¹) was calculated for bacteria, assuming exponential growth using:

$$GR_{\text{bacteria}} = \frac{k_{\text{bacteria}}}{\ln(2)} \quad (4)$$

Diatom volume and biomass estimation

Cell volume in each bottle was estimated as a cylinder from average cell length and width using:

$$V_{\text{diatom}} = \frac{\pi \times l_{\text{diatom}} \times d_{\text{diatom}}^2}{4} \quad (5)$$

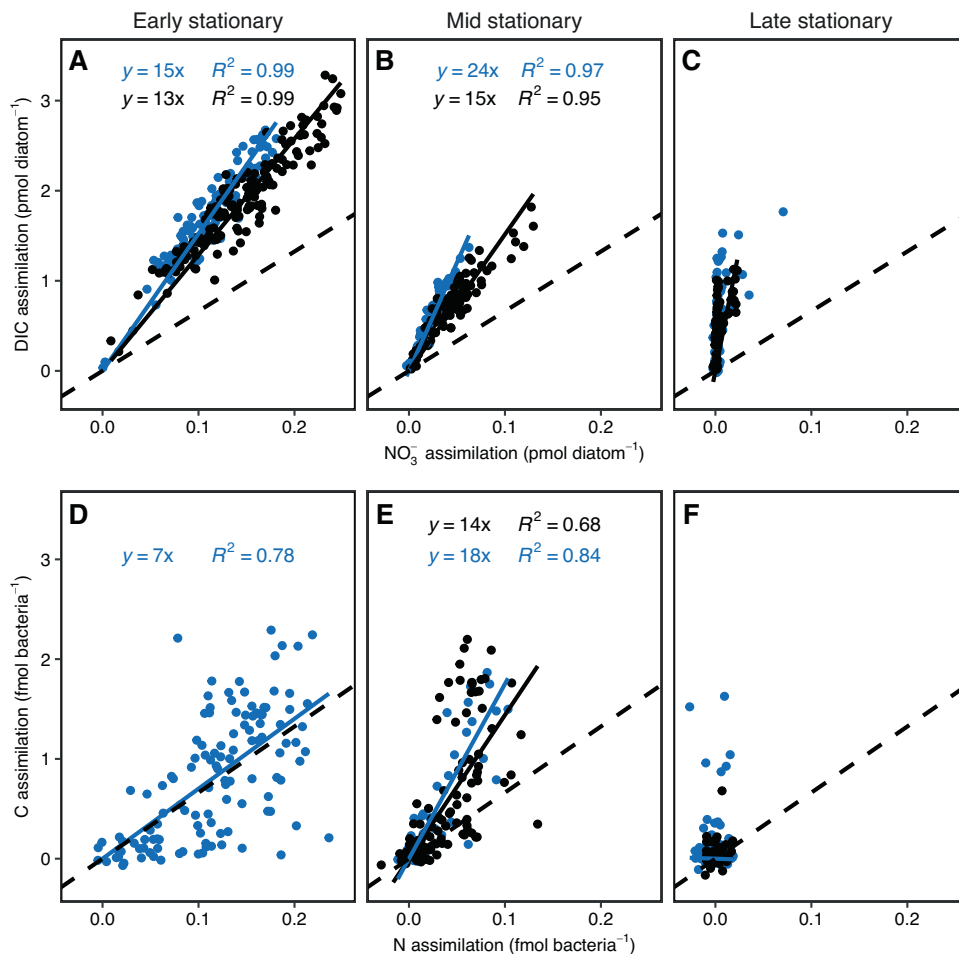


Fig. 1 C and N assimilation by diatoms and transfers to their attached bacteria. A–C DIC and NO₃⁻ assimilation in individual diatom cells, captured using SIMS, (D–F) Assimilation of freshly diatom derived C and N by individual diatom attached bacteria, captured using nanoSIMS. A, D Early stationary growth-phase, (B, E) mid stationary growth-phase and (C, F) late stationary growth-phase. Blue represents 12 h L and black 12:12 h L:D incubations. The dashed black line shows the Redfield C:N ratio and solid lines represent the observed C:N assimilation ratio. Note the different units between diatoms and bacteria (A–C and D–F).

V = diatom volume (μm^3), l = length (μm) and d = diameter (μm) [46]. Cell specific C content was calculated:

$$C_{\text{diatom}} = 0.288 \times V_{\text{diatom}}^{0.811} \quad (6)$$

$C_{\text{diatom}} = 10^{-12}$ g C diatom $^{-1}$ and V = total cell volume in μm^3 (ref. [47]). N_{diatom} was estimated from C_{diatom} using the Redfield ratio [48].

Bacterial volume and biomass estimation

Bacterial volume (μm^3) was estimated assuming a rod-shape approximated as a cylinder with a half-sphere at each end:

$$V_{\text{bacteria}} = \frac{4}{3} \times \pi \times r_{\text{bacteria}}^3 + \pi \times l_{\text{bacteria}} \times r_{\text{bacteria}}^2 \quad (7)$$

where l_{bacteria} is bacterial length and r_{bacteria} is bacterial radii (μm). C and N biomass (10^{-15} g bacteria $^{-1}$) was calculated from [49].

$$C_{\text{bacteria}} = 197 \times V_{\text{bacteria}}^{0.46} \quad (8)$$

$$N_{\text{bacteria}} = 39 \times V_{\text{bacteria}}^{0.38} \quad (9)$$

where C_{bacteria} and N_{bacteria} is the bacterial C and N biomass per bacteria (10^{-15} g).

Statistical analysis

We ensured that enough single cells were analyzed to achieve representative, stable mean values by plotting the cell-specific activity against number of cells analyzed [10]. Figures were produced using ggplot2 and the tidyverse package, error-bars represent SE [50]. Diatoms were considered active/enriched if their average C enrichment ratio minus two times the standard deviation was larger than the natural abundance ratio plus two times the standard deviation in the unlabeled control. Additionally, cells were visually inspected for bacterial remineralization. A chi-square test was used to investigate the number of active/inactive cells in chains and found as solitary cells. The level of significance was set to $p \leq 0.05$. We compared the observed distributions of active and inactive cells in chains to expected values in chains, calculated using probability theory, each chain was considered a Bernoulli process. Observed frequency of active and inactive cells was used to predict the theoretical distributions of fully active chains, mixed, or fully inactive chains when randomly

drafting cells for chains for each chain length. The expected probability of finding a fully active or in active chain (P) was calculated:

$$P = i^n \quad (10)$$

where, n is the chain length and i are the probability of finding an inactive or active cell. The odds of mixed chains were calculated by subtracting the odds of fully active and inactive chains from 1.

RESULTS

Diatom growth and dissolved nutrients

The *Chaetoceros* cultures reached the stationary phase after 9 days (Fig. S1). DIC was $890 \pm 27 \mu\text{M}$, $815 \pm 22 \mu\text{M}$, and $792 \pm 7 \mu\text{M}$ in the early (9 d), mid (15 d), and late stationary phase (21 d), respectively (Table S1). Dissolved NO_3^- decreased from, $42 \pm 6 \mu\text{M}$ during the early stationary phase to $39 \pm 9 \mu\text{M}$ and $28 \pm 5 \mu\text{M}$ in the mid and late stationary phase. Simultaneously, dissolved NH_4^+ increased from $0.20 \pm 0.02 \mu\text{M}$ (early) to $0.35 \pm 0.04 \mu\text{M}$ (mid), and $1.59 \pm 0.20 \mu\text{M}$ (late), representing 0.46%, 0.88%, and 5.37% of total DIN concentrations (Table S1). The silicate concentration decreased from $211 \pm 28 \mu\text{M}$ to $194 \pm 45 \mu\text{M}$, and $139 \pm 24 \mu\text{M}$ during the early, mid, and late stationary phase, respectively. Concentration of dissolved phosphate decreased from $8.9 \pm 0.3 \mu\text{M}$ to $6.7 \pm 1 \mu\text{M}$ and $8.0 \pm 0.2 \mu\text{M}$ (Table S1).

Cell-specific C- and N assimilation in diatoms

Despite high ambient NO_3^- concentrations, the average NO_3^- assimilation in diatom cells decreased from $114 \pm 4 \text{fmol diatom}^{-1}$ in the early stationary phase to $26 \pm 2 \text{fmol diatom}^{-1}$, and $4.7 \pm 1.2 \text{fmol diatom}^{-1}$ measured after 12 h L in the mid and late stationary phase, respectively (Fig. 1 and Table S2). The cell specific DIC: NO_3^- assimilation ratio was 15.1, 25.2, and 97.9 in the early, mid, and late stationary phases, respectively, and thus significantly higher than Redfield ratio of 6.6 (Table S3). The diatoms continued to assimilate NO_3^- during darkness at a low rate, but the overall C:N assimilation ratio after 12:12 h L:D cycle remained higher than the Redfield ratio. C and N assimilation did not result in net cellular

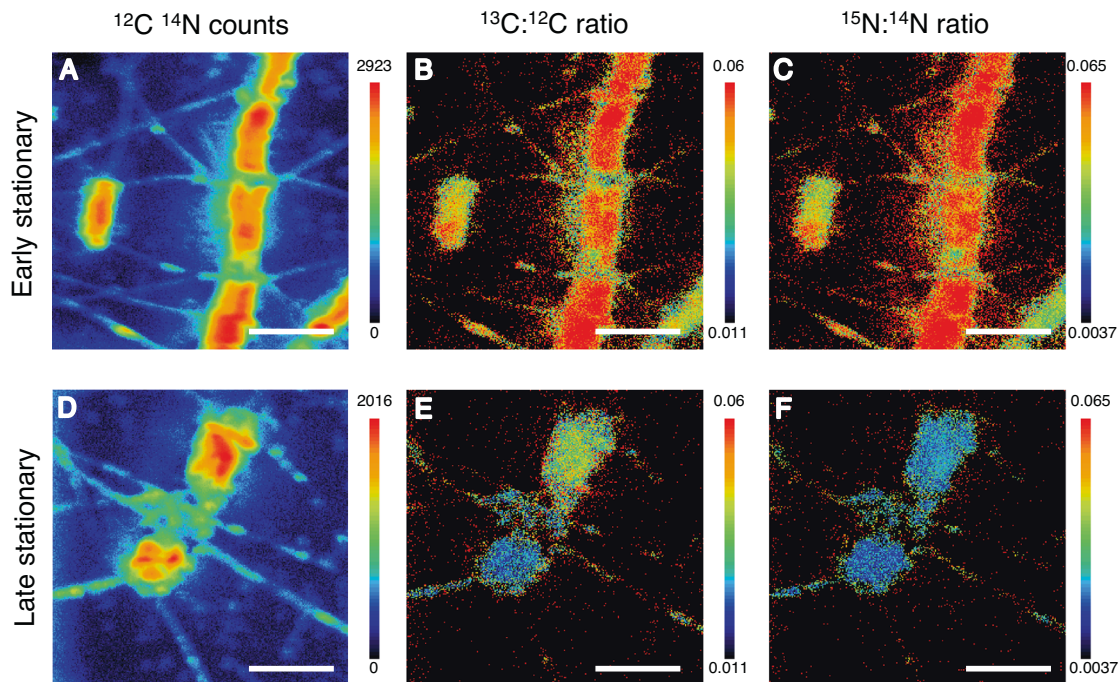


Fig. 2 DIC and NO_3^- assimilation in diatom chains and solitary cells captured using SIMS. **A–C** early stationary growth phase, **D–F** late stationary growth phase, **(A, D)** $^{12}\text{C}^{14}\text{N}$ ion counts per pixel, a proxy for diatom biomass used to determine the location of each cell. **(B, E)** $^{13}\text{C}^{12}\text{C}$ ratio, a proxy for DIC assimilation, note the higher variance between cells in subfigure (E) compared to (B). **C, F** $^{15}\text{N}^{14}\text{N}$ ratio, a proxy for NO_3^- assimilation. Scalebar represents 20 μm .

Table 1. Active and inactive diatom cells in chains and solitary cells during each growth phase, inactive cells were determined by visual inspection for bacterial remineralization or as having a 13C:12C ratio -SD x 2 smaller than that of the natural abundance + SD x 2.

Growth phase	Chain or solitary diatom	Active diatoms	Inactive diatoms	DIC assimilation active + inactive (pmol diatom ⁻¹)	DIC assimilation only active (pmol diatom ⁻¹)	NO ₃ ⁻ assimilation active + inactive (pmol diatom ⁻¹)	NO ₃ ⁻ assimilation only active (pmol diatom ⁻¹)
Early stationary	In chain	180 (98.4%)	2 (1.6%)	1.83 ± 0.04	1.85 ± 0.04	0.132 ± 0.004	0.133 ± 0.003
Early stationary	Solitary cell	76 (96.2%)	3 (3.8%)	1.83 ± 0.06	1.88 ± 0.05	0.130 ± 0.005	0.134 ± 0.005
Mid stationary	In chain	103 (85.8%)	17 (14.2%)	0.64 ± 0.03	0.719 ± 0.03 ^a	0.033 ± 0.002	0.037 ± 0.002 ^a
Mid stationary	Solitary cell	30 (75.0%)	10 (25.0%)	0.70 ± 0.07	0.867 ± 0.06 ^a	0.042 ± 0.005	0.054 ± 0.005 ^a
Late stationary	In chain	82 (47.4%)	91 (52.6%)	0.50 ± 0.04	0.714 ± 0.04	0.005 ± 0.001	0.007 ± 0.001
Late stationary	Solitary cell	35 (55.6%)	28 (44.4%)	0.44 ± 0.05	0.741 ± 0.06	0.005 ± 0.001	0.008 ± 0.002

The percentages represent the proportion active and inactive cells for each growth phase. Average DIC and NO₃⁻ ± SE, for all cells and only active cells.

^aIndicates $p < 0.05$ calculated using a two tailed unpaired students t test between solitary cells and cell in chains for each growth phase.

growth during the mid and late stationary phase. No significant differences in diatom abundance between the early and the late stationary phase was determined by t test ($27,000 \pm 2200$ diatom cells mL⁻¹ and $28,000 \pm 1900$ diatom cells mL⁻¹, $p = 0.70$) and cell size was stable (Tables S3 and S4). POC:PON mol ratio of the total biomass (diatoms and bacteria) was 7.0, 7.5, and 7.6 after 12:12 h L:D for the early, mid, and late stationary phases (Table S3), measured POC and PON per cell corresponded to values estimated from cell size (Table S5). In the early stationary phase, diatoms within the same chain showed similar C and NO₃⁻ assimilation rates independent of position in the chain (Figs. 2, S2 and S3). However, C and NO₃⁻ assimilation varied between chains, although the C:N ratio was stable within replicates. Chain-length decreased from 5.20 ± 0.13 diatom cells chain⁻¹ in the early stationary phase to 2.56 ± 0.08 diatom cells chain⁻¹ in the late stationary phase (Table S4), concurrent with increasing variability of C assimilation and a decrease in NO₃⁻ assimilation rates within the same chain (Fig. 1 and Tables 1 and S6).

Bacteria-diatom interactions

Diatom attached bacteria increased during the stationary phase. Each diatom harbored on average 10 ± 0.4 bacteria diatom⁻¹, 36 ± 4 bacteria diatom⁻¹, and 86 ± 4 bacteria diatom⁻¹ in the early, mid, and late stationary phase, respectively. The average exponential growth rate of attached bacteria was 0.0075 h⁻¹ (Fig. S4 and Table S2). No autofluorescence was observed in the diatom attached bacteria, which were assumed to be heterotrophs, consuming dissolved organic matter, and releasing NH₄⁺. This was supported by the observation that bacteria assimilated both C and N derived from diatoms close to Redfield ratio during the early stationary phase (Fig. 1). Diatom attached bacteria assimilated 0.82 ± 0.06 fmol freshly diatom-derived C bacteria⁻¹, 0.45 ± 0.07 fmol C bacteria⁻¹, and 0.11 ± 0.03 C bacteria⁻¹ during the early, mid, and late stationary phase, after 12:12 h L:D. During the same period, diatom attached bacteria also assimilated 0.116 ± 0.005 fmol freshly diatom-derived N bacteria⁻¹, 0.023 ± 0.004 fmol N bacteria⁻¹, and 0.000 ± 0.001 fmol N bacteria⁻¹ (Fig. 1, and Table S2). Free-living bacteria assimilated 0.17 ± 0.05 fmol freshly diatom-derived C bacteria⁻¹, 0.13 ± 0.03 fmol C bacteria⁻¹, and 0.07 ± 0.02 fmol C bacteria⁻¹ during the early, mid, and late stationary phases after 12:12 h L:D and 0.084 ± 0.015 fmol freshly diatom derived N bacteria⁻¹, 0.009 ± 0.002 fmol N bacteria⁻¹, and -0.002 ± 0.001 fmol N bacteria⁻¹. Hence, diatom attached bacteria assimilated up to 4.7 times more C during the early stationary phase and up to 2.6 times more N freshly derived from diatom DIC and DIN assimilation than their free-living counterparts during the mid-stationary phase. Up to 3.2 and 5.1% of freshly assimilated ¹³C and ¹⁵N was detected in attached bacteria within 12 h and supported bacterial C- and N- specific growth rates up to 0.026 h⁻¹ and 0.008 h⁻¹, respectively, close to the growth rate observed with DAPI of 0.0075 h⁻¹ (Table S7).

C and N dynamics in chains and solitary diatom cells

Average DIC and NO₃⁻ assimilation rates decreased simultaneously for both solitary cells and chain cells over the stationary phase (Fig. 1 and Table 1 and S6). On average, chain length decreased by 31% while C and NO₃⁻ assimilation in chain cells decreased by 65 and 75%, from the early to the mid stationary phase respectively (Table S6). Hence, C and NO₃⁻ assimilation in chain cells on average decreased comparatively more than chain length from the early to the mid stationary phase. However, statistical analysis of nutrient assimilation as a function of chain length under similar growth conditions was not feasible because short cell chains were rare in the early stationary phase, whereas long chains were rare in the mid and late stationary phase. Cell abundance remained relatively stable while chain length decreased (Tables S3, S4). Hence, long chains split into shorter chains during the stationary phase. Furthermore, cells with high C assimilation rates occurred frequently adjacent to inactive cells in the same chain during the late stationary phase

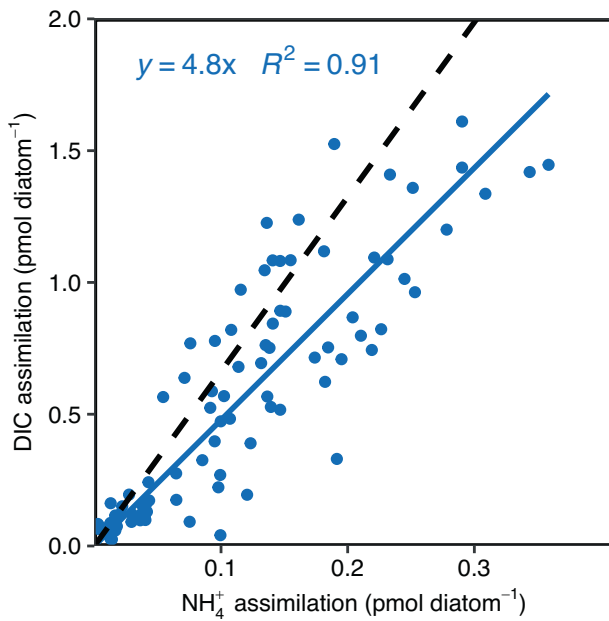


Fig. 3 DIC and NH_4^+ assimilation by diatoms measured using SIMS extrapolated to 12 h L from 4.15 h L during the late stationary phase. The dashed line represents Redfield C:N ratio and the solid line represent the average observed C:N assimilation ratio.

(Figs. 2 and S3). The proportion of active chain cells declined from 98% in the early stationary phase to 47% in the late stationary phase. By comparison, the fraction of active solitary diatom cells declined from 96 to 56% during the same time (Table 1). The proportion of active/inactive cells within chains and solitary cells were not significantly different at any timepoint, as determined by a chi-square test ($p = 0.53, 0.18, \text{ and } 0.34$) for the early mid and late stationary phase). The diatom cell-specific DIC, NO_3^- , and NH_4^+ assimilation rates were not significantly different between cells, within chains, and solitary cells due to high variances when active and inactive cells were pooled. However, both C and NO_3^- assimilation by solitary cells were significantly higher as compared to cells in chains when only active cells were considered during the mid-stationary phase (Table 1). In the late stationary growth phase, bacterially regenerated NH_4^+ was the main N source for diatoms despite a low ambient concentration (5.4% of available DIN) compared to NO_3^- (Table S1). The NH_4^+ assimilation rate was $110.5 \pm 9.4 \text{ fmol diatom}^{-1}$ and exceeded the NO_3^- assimilation rates measured in the early stationary phase, contributing 96% of the diatoms DIN assimilation in the late stationary phase. The DIC: NH_4^+ assimilation ratio during the late stationary phase was close to the Redfield ratio (6.6) in contrast to the DIC: NO_3^- assimilation ratio of 97.9 (Figs. 1 and 3).

Distributions of active and inactive chain cells

Chain length decreased during the stationary phase leading to chains with two to four cells, only. Two-cell chains represented 33% of all cells in chains (58 of 173 observations). Of these two-cell chains, 51.7% (15 of 29 observations) were comprised of one active and one inactive cell while 24.1% (7 of 29 observations) were comprised of two active or inactive cells (Table 2). Three-cell chains represented 36.4% of all chain cells (63 of 173 observations) and represented the highest fraction of active cells (58.7%). Of all three-cell chains, 75% contained both active and inactive cells. Four-cell chains represented 30% of the cells in chains (52 of 173 observations). One third (33.3%) of the four cell chains were completely inactive, and 66.7% of the chains contained a mixture of active and inactive cells (Table 2). The distributions of active and

Table 2. Active and inactive diatoms in chains of varying length during the late stationary growth phase.

Chain length	Active cells	Inactive cells	Fully active chains	Mixed chains	Fully inactive chains	Expected fully active chains	Expected mixed chains	Expected fully inactive chains
Two-cell chain	29 (50%)	29 (50%)	7 (24.1%)	15 (51.7%)	7 (24.1%)	7 (22.5%)	14 (49.9%)	8 (27.7%)
Three-cell chain	37 (58.7%)	26 (41.3%)	3 (15.0%)	15 (75.0%)	2 (10.0%)	2 (10.6%)	16 (74.8%)	3 (14.6%)
Four-cell chain	16 (30.8%)	36 (69.2%)	0 (0.0%)	8 (66.7%)	4 (33.3%)	1 (5.0%)	11 (87.3%)	1 (7.7%)
Total	82 (47.4%)	91 (52.6%)						

The bold numbers are the sum of observations above and the relative frequency of active and inactive cells, used to calculate the probability of active and inactive cells in chains of different lengths. Cells was checked for significant C assimilation (see method or Table 1 for more details) and visually for bacterial degradation. Expected values for chains were calculated using probability theory. The total chance for an active (47.4%) or inactive cell (52.6%) was used in a sequential biased coinflips to predict the chance of getting a fully active, mixed or fully inactive chain when randomly drafting cells for chains of length.

inactive cells in chains corresponded well to those predicted by probability theory for two- and three-cell chains. Four-cell chains had a higher portion of completely inactive chains than predicted and the number of inactive cells was significantly higher compared to the three-cell chains using a chi-square test ($p = 0.005$).

DISCUSSION

SIMS was first used in diatom ecology to quantify Si/C assimilation ratios in individual diatoms [51]. Recently, cell specific C and N assimilation have been analyzed with respect to chain length, cell position within long chains, and degree of ambient water turbulence in natural field populations [6, 10, 11]. Studies using nanoSIMS have focused on the microbial activities in the diatom phycosphere and in diatom aggregates [14, 15, 24]. Earlier studies demonstrated that both diatoms and bacteria can experience increased growth and survival during the exponential as well as the stationary phase during co-existence in cultures as compared to axenic diatom culture and bacterial mono cultures [27]. A host of synergistic interactions could explain this, see [22] for a review. Here, we quantitatively studied the dynamics and mutual benefits of interactions between diatoms and attached bacteria within diatom chains and solitary cells. We focused on the stationary phase characterized by complex diatom dynamics considering chain length and nutrients, and interactions with attached bacteria to gain a mechanistic understanding of the C and N exchange associated with diatom chains and solitary cells. To our knowledge, this is the first study to directly quantify cell-to-cell variations and dynamics of N assimilation and transfer within diatom chain cells as consequences of microbial activities.

Variance of DIC and N assimilation within diatom chains

Diatom chains form during cell division, when the silica tubule between daughter cells are fused together [52]. This results in chains consisting solely of clones. Chain length decreases if daughter cells with terminal setae are produced, which splits the chain during division [53, 54]. During mitosis, diatoms produce one daughter cell of the original cell size and one slightly smaller, leading to a decrease in average cell size. Cell volume can vary up to an order of magnitude between the largest and smallest cells of the same clonal strain [55]. Our study shows that while DIC and NO_3^- assimilation rates varied between chains, the within-chain variation was low during the early stationary phase, likely due to the clonal nature of diatom chains. Cell-specific C and NO_3^- assimilation rates varied 2- to 3-fold among all cells in chains during the early

stationary phase similar to those previously reported for the chain forming diatom, *Skeletonema*, during the exponential phase [11]. During the mid and late stationary phases, we observed an increasing proportion of inactive diatoms both within chains and as solitary cells (Table 1). A significantly higher assimilation of both DIC and NO_3^- occurred in solitary cells compared to cells in chains during the mid-stationary growth phase when only active cells were considered. Thus, the presence of inactive cells masked the signal of active cells. The majority of highly active cells in chains of all lengths (2–4 diatom cells chain^{-1}) occurred in chains with at least one inactive cell during the late stationary phase, as predicted by probability theory (Table 2). In two-cell chains (Fig. 2), both cells have the same potential to assimilate nutrients from the ambient water according to diffusion theory. However, we observed that one cell was active while the other was inactive in over half of the chains. Such an imbalance rules out diffusion limitation from the ambient water in two-cell chains and may even suggest transfer of nutrients to active cells from neighboring inactive cells. Diffusion time of ions and solutes increases with the square of distance [56]. Owing to the short diffusion distance between neighboring chain cells, NH_4^+ remineralized by bacteria from dead diatoms may be efficiently transferred to living, neighboring cells by diffusion (Fig. S5). This NH_4^+ source would have had a natural ^{15}N ratio. Our method therefore underestimates NH_4^+ -based primary production by neighboring cells if this unlabeled NH_4^+ diffuses to a neighboring cell before mixing with the ^{15}N -enriched ambient water (Fig. 4). Our cultures had higher cell abundances ($\sim 30,000$ cells mL^{-1}) than in nature and the NH_4^+ produced from remineralized cells in the ambient water was high and could balance C fixation. In a low N regime with competing phytoplankton, the short diffusion distance between neighboring cells may significantly increase N transfer for surviving diatom cells in the same chain. It remains challenging to design experiments that directly measure these fluxes with methodologies available today due to the small spatial scales these fluxes operate on.

Bacterial interactions

Diatoms supported substantial growth of attached and free-living bacteria by excreting freshly (<12 h old) assimilated and transformed DIC and DIN to the immediate environment (Table S2), presumably as dissolved organic matter. Such diatom-derived exudates can provide all nutrients required for bacterial growth [57]. Diatom attached bacteria clearly benefitted more from this release from their attachment directly to the source, as opposed to their free-living counterparts (Table S2). In return, diatoms could later

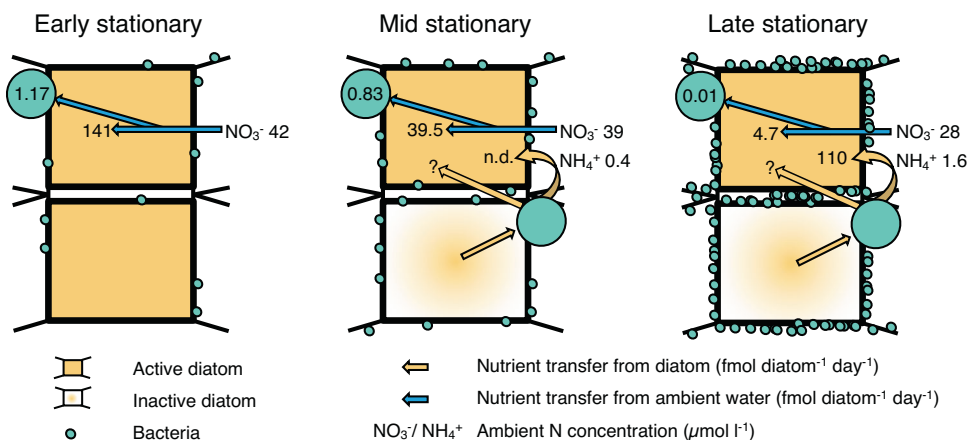


Fig. 4 Measured N transfer between active diatoms, diatom attached bacteria, and inactive diatoms within a chain, during the early, mid and late stationary phase. Note how N assimilation shifts from NO_3^- in the ambient water to NH_4^+ originating from bacterially remineralized neighboring diatoms between the growth phases. The question marks represent the 'masked' flux of N from a bacterially remineralized diatom directly to the neighbor, before mixing with the ambient ^{15}N - NH_4^+ enriched water. See Tables S1 and S2 for more details and SE.

balance their primary production with NH_4^+ assimilation from bacterial remineralization. The diatom $\text{DIC}:\text{NH}_4^+$ assimilation ratio was 4.8 during 12 h L (Fig. 3). Thus, bacteria supported diatoms with remineralized NH_4^+ at a rate which was close to that of NO_3^- assimilation during the early stationary phase, as previously suggested by Grossart [27]. In the early stationary phase, nutrients transferred to attached bacteria did not exceed 0.44 and 0.83% of total diatom C and N assimilation, respectively (Table S2). Thus, “microbial gardening” of bacteria serves quantitatively as a low-cost solution to diatoms as an alternative NH_4^+ source to support living diatoms and completion of the life cycle after the late stationary phase (see ‘introduction’). NH_4^+ production from the random distribution of decay within diatom chains offers a previously overlooked nutrient source for surviving cells in the same chain as well as a mechanism to reduce the number of clones before sexual reproduction. Hence, chain formation and microbial interactions support both short term and long survival in diatoms.

DATA AVAILABILITY

The datasets generated during and analyzed during the current study are available from the corresponding author upon reasonable request.

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AUTHOR CONTRIBUTIONS

HP, EKR, and RS conceived and designed the experiment, RS and EKR performed the experiments. RS, EKR, and MJW analyzed the samples, RS, HP, and EKR analyzed the data and wrote the paper. All authors have reviewed and accepted the paper.

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COMPETING INTERESTS

The authors declare no competing interests.

ADDITIONAL INFORMATION

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